-Communications to the editor

The Synthesis of Racemic Allococaïne and Racemic Allopseudococaïne

Sir:

Willstätter and his collaborators demonstrated that cocaïne, one of the classical topics of alkaloid chemistry, is a 2-carbomethoxy-3-benzoxytropane and that the latter should exist as four stereoisomeric racemic (and eight optically active) modifications.^{1,2,3} It appears that, of these, only racemic cocaïne⁴ and racemic pseudococaïne have been synthesized^{2,3,4} and that only the former has been resolved.⁴ I wish to report that the other two possible racemic cocaïnes have recently been synthesized in this laboratory.

Hydrogenated in aqueous acetic acid with Adams' catalyst, racemic 2-carbomethoxytropinone^{4,5} furnishes in about 80% yield a third racemic ecgonine methyl ester (I) [Anal. Cale'd for C₁₀H₁₇NO₃: C, 60.3; H, 8.6. Found: C, 60.4; H, 8.8], m.p. 81.5-83.5°, which on benzoylation affords a third racemic cocaïne (II) [Anal. Calc'd for C₁₇H₂₁NO₄: C, 67.3; H, 7.0. Found: C, 67.3; H, 6.9], m.p. 82-84°. Hydrolysis of I gives a mixture of two racemic ecgonines, one of which melts at 242° (III) [Anal. Calc'd for C₉H₁₅NO₃: C, 58.4; H, 7.8. Found: C, 58.1; H, 8.0] (hydrochloride, m.p. 213°) and the other at 237° (IV). Esterification of the former compound results in the fourth racemic ecgonine methyl ester (V) [Anal. Calc'd for C10H17NO3: C, 60.3; H, 8.6. Found: C, 60.0; H, 8.6], m.p. 80°, which on benzoylation gives the fourth racemic cocaïne (VI) [Anal. Calc'd for C1:H21NO4: C, 67.3; H, 7.0. Found: C, 67.5; H, 6.7], m.p. 98°. The latter ecgonine (IV) gives a hydrochloride [Anal. Calc'd for $C_{9}H_{16}CINO_{3}$: C, 48.8; H, 7.3. Found: C, 49.0; H, 7.2] melting at 231–233° (dec.) and hence appears to be identical with Willstätter's 'drittes racemisches Ekgonin.'4 The picrates of I, II, V, and VI melt at 204°, 179°, 136°, and 162°, respectively.

Inasmuch as the two ecgonine methyl esters already known both have the β -configuration of the C₃-OH,^{6,7} the new isomers (I and V) must both have the α -configuration. In its reaction with methyl iodide, V resembles ecgonine methyl ester and I pseudoecgonine methyl ester.⁸ For reasons given earlier,⁷ this indicates that V has the β configuration of the 2-carbomethoxy group. Accordingly, it is tentatively concluded that II is allococaïne and VI allopseudococaïne.⁷

2,4-Dicarbomethoxytropinone, prepared by Robinson's biological method⁹ (binoxalate, m.p. 148°), is saponifiable to racemic 2-carbomethoxytropinone.^{4,5} Preliminary experiments indicate that *d*-pseudoecgonine methyl ester can feasibly be oxidized with chromic-sulfuric acid in acetone to the optically active form of this β -keto ester. Hence it appears that all the cocaïnes, optically active as well as racemic, are obtainable and their relative and absolute configurations ascertainable.¹⁰

I hope to give soon a detailed account of the preparation, properties, stereochemistry, resolution, and pharmacological properties of certain of the foregoing and related compounds.



(8) Cf., reference 4, pp. 127-128.

(9) Cf., Robinson, J. Chem. Soc., 111, 762 (1917); see also Schöpf and Lehmann, Ann., 518, 1 (1935).

(10) Cf., Hardeggar and Ott, Helv. Chim. Acta, 38, 312 (1955).

Received April 9, 1956

The ortho-Alkylation of Aromatic Amines

Sir:

The nuclear alkylation of aromatic amines in the past has not been regarded as a practical reaction because of complications arising from the characteristics of the amino group. We wish to report a new reaction by which primary and secondary aromatic amines are alkylated with olefins exclusively in the *ortho* positions. The reaction is of interest both from a theoretical viewpoint and also in that it provides a direct route to

⁽¹⁾ Willstätter and Müller, Ber., 31, 2655 (1898).

⁽²⁾ Willstätter and Bode, Ann., 326, 42 (1903).

⁽³⁾ Willstätter and Bommer, Ann., 422, 15 (1921).

⁽⁴⁾ Willstätter, Wolfes, and Mäder, Ann., 434, 111 (1923).

⁽⁵⁾ Preobrashenski, Schtschukina, and Lapina, Ber., 69, 1615 (1936).

⁽⁶⁾ Fodor and Kovács, J. Chem. Soc., 724 (1952).

⁽⁷⁾ Findlay, J. Am. Chem. Soc., 75, 1033 (1953); 76, 2855 (1954).